

Name: _____

Score: _____ Date: _____

Molecular and Ionic Equations

Write down the complete molecular, ionic, & net ionic equations for the following.

1) Potassium phosphate and calcium nitrate

Molecular equation:

Ionic equation:

Net ionic equation:

2) Potassium hydroxide and copper(II) nitrate

Molecular:

Ionic:

Net ionic:

3) Calcium iodide and potassium carbonate

Molecular:

Ionic:

Net ionic:

4) Barium nitrate and potassium sulfate

Molecular:

Ionic:

Net ionic:

5) Lead(II) perchlorate and sodium chloride

Molecular:

Ionic:

Net ionic:

6) Silver nitrate and magnesium iodide

Molecular:

Ionic:

Net ionic:

7) Potassium fluoride and magnesium nitrate

Molecular:

Ionic:

Net ionic:

8) Hydrobromic acid and lead(II) perchlorate

Molecular:

Ionic:

Net ionic:

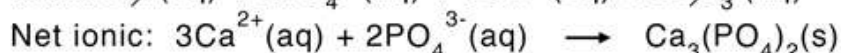
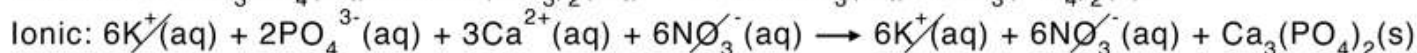
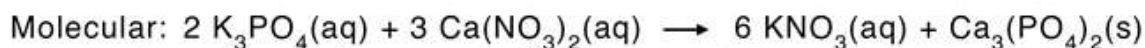
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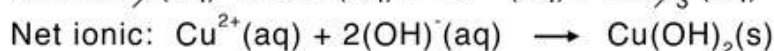
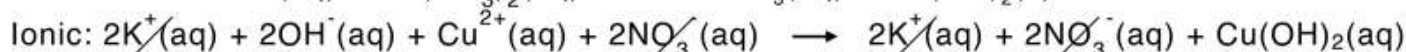
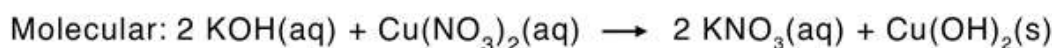
Molecular and Ionic Equations (Answers)

Write down the complete molecular, ionic, & net ionic equations for the following.

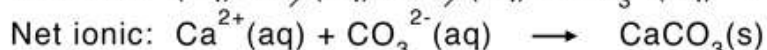
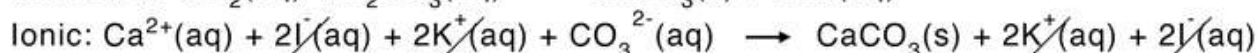
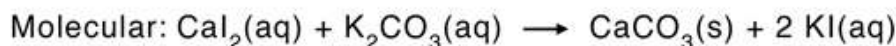
1) Potassium phosphate and calcium nitrate



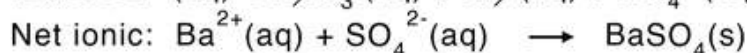
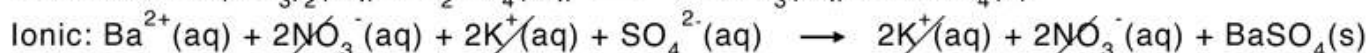
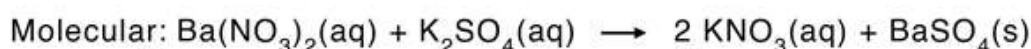
2) Potassium hydroxide and copper(II) nitrate



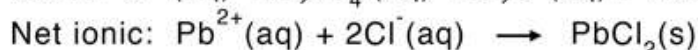
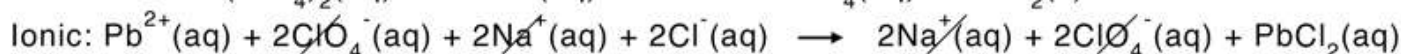
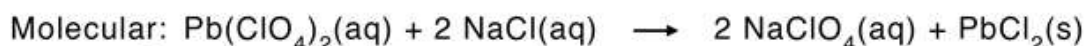
3) Calcium iodide and potassium carbonate



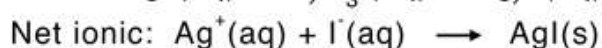
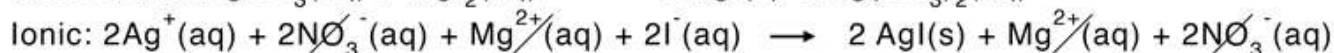
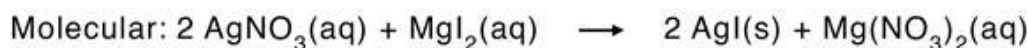
4) Barium nitrate and potassium sulfate



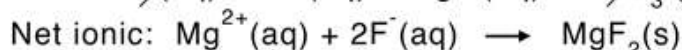
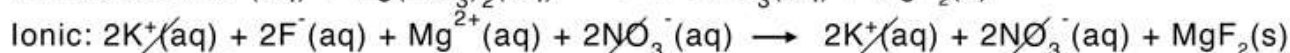
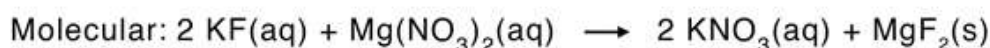
5) Lead(II) perchlorate and sodium chloride



6) Silver nitrate and magnesium iodide



7) Potassium fluoride and magnesium nitrate



8) Hydrobromic acid and lead(II) perchlorate

