| Nam            | e:  |  |                              | Date :            |     |  |  |  |  |
|----------------|---|--|------------------------------|-------------------|-----|--|--|--|--|
|                |   |  | Workshee                     |                   |     |  |  |  |  |
| (1) V          | Vhat are isotopes?  |  |                              |                   |     |  |  |  |  |
|                |   | pe is written in wo                                      | ord, a dash, and a n<br>ify? | umber, what does  | the |  |  |  |  |
| ( <u>3</u> ) F | How can you tell isotopes apart?  |  |                              |                   |     |  |  |  |  |
| ( <u>4</u> ) C | Complete the table for the following isotopes.  |  |                              |                   |     |  |  |  |  |
|                |   |  | Iodine - 125                 | Iodine - 128      |     |  |  |  |  |
|                |   | # of protons   |                              |                   |     |  |  |  |  |
|                |   | # of neutrons  |                              |                   |     |  |  |  |  |
|                |   | # of electrons   |                              |                   |     |  |  |  |  |
|                |   |  | Chromium - 58                | lodine - 125      |     |  |  |  |  |
|                |   | # of protons   |                              |                   |     |  |  |  |  |
|                |   | # of neutrons  |                              |                   |     |  |  |  |  |
|                |   | # of electrons   |                              |                   |     |  |  |  |  |
| <u>(5)</u> (   | ircle the corr  | ect answer for the                                       | following multiple           | -choice auestions | I   |  |  |  |  |
| i)             | ircle the correct answer for the following multiple-choice questions.  Fluorine is represented by which of the following chemical symbol? |  |                              |                   |     |  |  |  |  |
|                | a) F  | b) Fe  | c) Fi                        | , d)              |     |  |  |  |  |
| ii)            | The compou  | ınd C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> contai | s of oxygen?                 |                   |     |  |  |  |  |
|                | a) 4  | b) 6   | c) 12                        | d)                | 14  |  |  |  |  |
| iii)           | i) An atom which has 9 protons, 8 neutrons, and 9 electrons will have an atomic mass of how many amu?                                     |  |                              |                   |     |  |  |  |  |
|                | a) 6  | b) 8   | c) 9                         | d)                | 17  |  |  |  |  |
| iv)            | An atom of r  | nagnesium has ho   | ow many protons?             |                   |     |  |  |  |  |
|                | a) 12   | b) 20  | c) 24                        | d)                | 25  |  |  |  |  |

| Name :   |   |                                 |                |               | Date :       |    |  |  |  |
|--|---|---------------------------------|----------------|---------------|--------------|----|--|--|--|
| A  | t   |                                 |                | Workshee      |              |    |  |  |  |
| (1)  | W   | /hat are isoto                  | pes?           |               |              |    |  |  |  |
| Atoms having the same number of protons but different number of neutrons |   |                                 |                |               |              |    |  |  |  |
| (2)  | When an isotope is written in word, a dash, and a number, what does the number next to the isotope signify? |                                 |                |               |              |    |  |  |  |
|  | Т   | The mass number of the isotope. |                |               |              |    |  |  |  |
| (3)  | How can you tell isotopes apart?  |                                 |                |               |              |    |  |  |  |
|  | From the number of neutrons or from the mass number.  |                                 |                |               |              |    |  |  |  |
| <u>[4]</u>   | Complete the table for the following isotopes.  |                                 |                |               |              |    |  |  |  |
|  |   |                                 |                | lodine - 125  | lodine - 128 |    |  |  |  |
|  |   |                                 | # of protons   | 53            | 53           |    |  |  |  |
|  |   |                                 | # of neutrons  | 72            | 75           |    |  |  |  |
|  |   |                                 | # of electrons | 53            | 53           |    |  |  |  |
|  |   |                                 |                | Chromium - 58 | lodine - 125 |    |  |  |  |
|  |   |                                 | # of protons   | 24            | 24           |    |  |  |  |
|  |   |                                 | # of neutrons  | 34            | 39           |    |  |  |  |
|  |   |                                 | # of electrons | 24            | 24           |    |  |  |  |
| © Circle the correct answer for the following multiple-choice questions. |   |                                 |                |               |              |    |  |  |  |
|  | i) Fluorine is represented by which of the following chemical symbol?                                       |                                 |                |               |              |    |  |  |  |
|  |   | a) F                            | b) Fe          | c) Fi         | d)           | FI |  |  |  |
|  | ii) The compound C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> contains how many atoms of oxygen?           |                                 |                |               |              |    |  |  |  |
|  |   | a) 4                            | (b) 6          | c) 12         | d)           | 14 |  |  |  |
|  | iii) An atom which has 9 protons, 8 neutrons, and 9 electrons will have an atomic mass of how many amu?     |                                 |                |               |              |    |  |  |  |
|  |   | a) 6                            | b) 8           | c) 9          | (d)          | 17 |  |  |  |
|  | iv) An atom of magnesium has how many protons?  |                                 |                |               |              |    |  |  |  |

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c) 24

b) 20

d) 25