

Writing Electron Configuration & Orbital Diagram

1. Write the long-form electron configuration of the following elements?

i. Lithium

ii. Magnesium

iii. Potassium

iv. Nickel

v. Bromine

2. Write the abbreviated electron configuration of the following elements?

i. Nitrogen

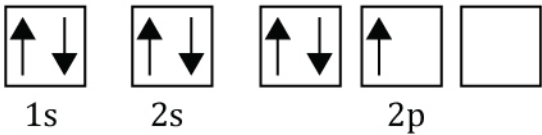
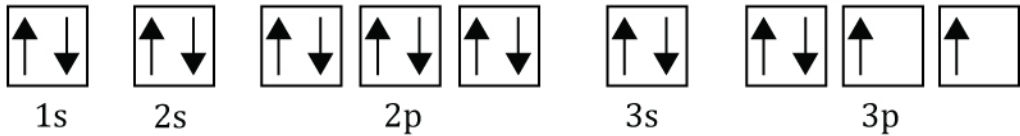
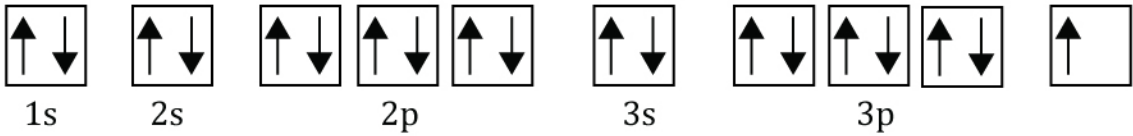
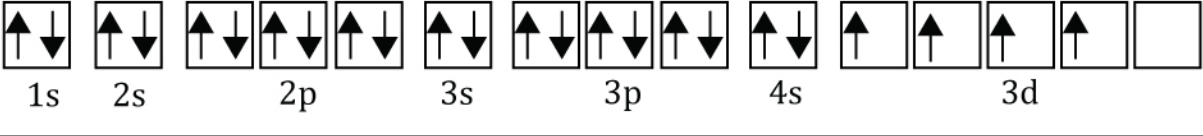
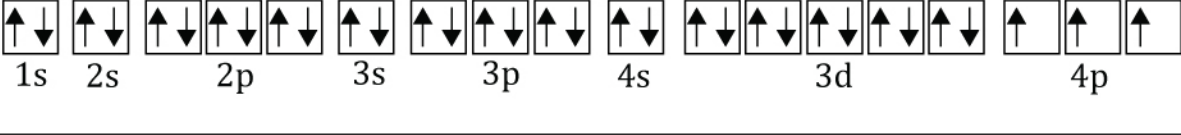
ii. Argon

iii. Manganese

iv. Silver

v. Barium

3. Identify the elements from the orbital filling diagrams.

Orbital Filling Diagram	Element
 <p style="margin-left: 100px;">1s 2s 2p</p>	
 <p style="margin-left: 100px;">1s 2s 2p 3s 3p</p>	
 <p style="margin-left: 100px;">1s 2s 2p 3s 3p</p>	
 <p style="margin-left: 100px;">1s 2s 2p 3s 3p 4s 3d</p>	
 <p style="margin-left: 100px;">1s 2s 2p 3s 3p 4s 3d 4p</p>	

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Answers

1. Write the long-form electron configuration of the following elements?

i. Lithium	$1s^2 2s^1$
ii. Magnesium	$1s^2 2s^2 2p^6 3s^2$
iii. Potassium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$
iv. Nickel	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^8$
v. Bromine	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$

2. Write the abbreviated electron configuration of the following elements?

i. Nitrogen	$[\text{He}] 2s^2 2p^3$
ii. Argon	$[\text{Ne}] 3s^2 3p^6$
iii. Manganese	$[\text{Ar}] 4s^2 3d^5$
iv. Silver	$[\text{Kr}] 5s^1 4d^{10}$
v. Barium	$[\text{Xe}] 6s^2$

3. Identify the elements from the orbital filling diagrams.

Orbital Filling Diagram	Element
<p style="margin: 5px 0;">$1s$ $2s$ $2p$</p>	C
<p style="margin: 5px 0;">$1s$ $2s$ $2p$ $3s$ $3p$</p>	S
<p style="margin: 5px 0;">$1s$ $2s$ $2p$ $3s$ $3p$</p>	K
<p style="margin: 5px 0;">$1s$ $2s$ $2p$ $3s$ $3p$ $4s$ $3d$</p>	Sr
<p style="margin: 5px 0;">$1s$ $2s$ $2p$ $3s$ $3p$ $4s$ $3d$ $4p$</p>	As