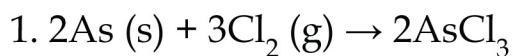


Name : \_\_\_\_\_ Date : \_\_\_\_\_

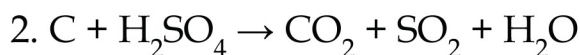
## Oxidation and Reduction Equations Worksheet

Determine the species being oxidized and reduced in the following equations.



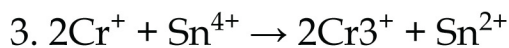
Oxidized:

Reduced:



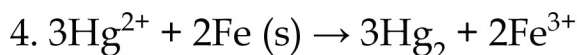
Oxidized:

Reduced:



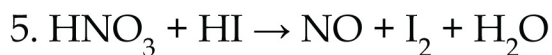
Oxidized:

Reduced:



Oxidized:

Reduced:



Oxidized:

Reduced:



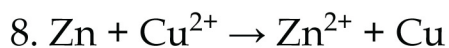
Oxidized:

Reduced:



Oxidized:

Reduced:



Oxidized:

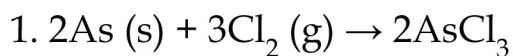
Reduced:

Name : ..... Date : .....

## Oxidation and Reduction Equations Worksheet

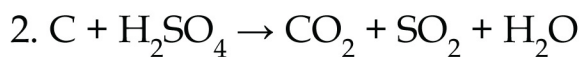
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### Answers



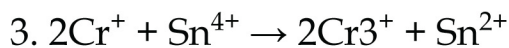
Oxidized: **As**

Reduced: **Cl<sub>2</sub>**



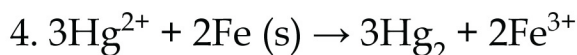
Oxidized: **C**

Reduced: **H<sub>2</sub>SO<sub>4</sub>**



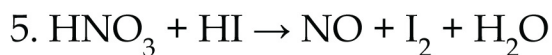
Oxidized: **Cr<sup>+</sup>**

Reduced: **Sn<sup>4+</sup>**



Oxidized: **Fe**

Reduced: **Hg<sup>2+</sup>**



Oxidized: **HI**

Reduced: **HNO<sub>3</sub>**



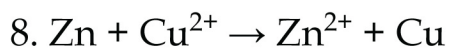
Oxidized: **HCl**

Reduced: **KMnO<sub>4</sub>**



Oxidized: **Sb**

Reduced: **HNO<sub>3</sub>**



Oxidized: **Zn**

Reduced: **Cu<sup>2+</sup>**