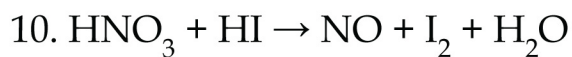
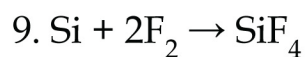
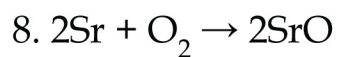
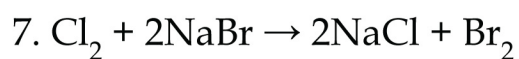
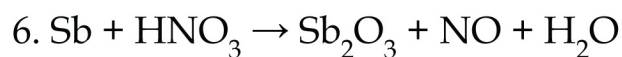
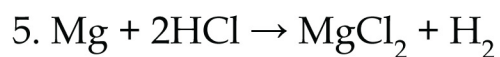
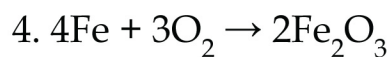
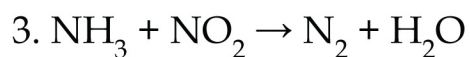
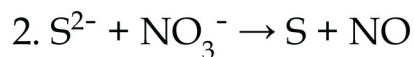
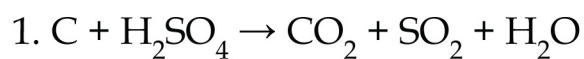


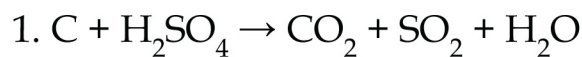
OXIDATION Worksheet

Determine in the following which atoms have been oxidized.

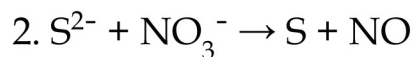


OXIDATION Worksheet

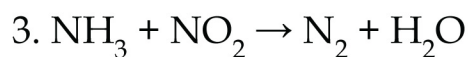
Answers



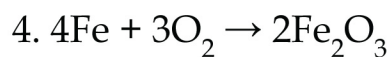
C is oxidized to CO_2 ; $0 \rightarrow +4$



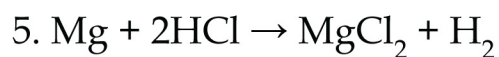
S^{2-} is oxidized to S; $-2 \rightarrow 0$



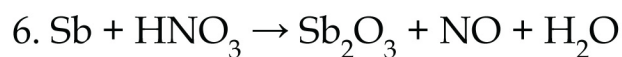
N^{3-} is oxidized to N; $-3 \rightarrow 0$



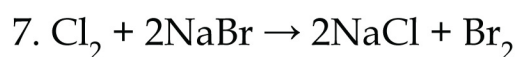
Fe is oxidized to Fe^{3+} ; $0 \rightarrow +3$



Mg is oxidized to Mg^{2+} ; $0 \rightarrow +2$



Sb is oxidized to Sb^{3+} ; $0 \rightarrow +3$



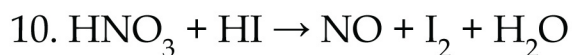
Br^- is oxidized to Br; $-1 \rightarrow 0$



Sr is oxidized to Sr^{2+} ; $0 \rightarrow +2$



Si is oxidized to Si^{4+} ; $0 \rightarrow +4$



I^- is oxidized to I; $-1 \rightarrow 0$