

# Periodic Table - Protons, Neutrons, and Electrons

Determine the number of protons, neutrons, and electrons for each element or ion. Write the atomic symbol for the element or ion as well.

<b>Hydrogen</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Carbon</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Oxygen</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Sodium ion</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Cholrine ion</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Magnesium</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Nitrogen</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

<b>Silver ion</b>	
Protons	
Neutrons	
Electrons	
Atomic Symbol	

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## Answers

<b>Hydrogen</b>	
Protons	1
Neutrons	0
Electrons	1
Atomic Symbol	H

<b>Carbon</b>	
Protons	6
Neutrons	6
Electrons	6
Atomic Symbol	C

<b>Oxygen</b>	
Protons	8
Neutrons	8
Electrons	8
Atomic Symbol	O

<b>Sodium ion</b>	
Protons	11
Neutrons	12
Electrons	10
Atomic Symbol	$\text{Na}^+$

<b>Cholrine ion</b>	
Protons	17
Neutrons	18
Electrons	18
Atomic Symbol	$\text{Cl}^-$

<b>Magnesium</b>	
Protons	12
Neutrons	12
Electrons	12
Atomic Symbol	Mg

<b>Nitrogen</b>	
Protons	7
Neutrons	7
Electrons	7
Atomic Symbol	N

<b>Silver ion</b>	
Protons	47
Neutrons	61
Electrons	46
Atomic Symbol	$\text{Ag}^+$