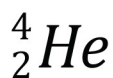
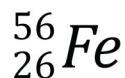
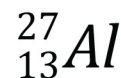
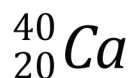
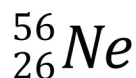
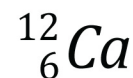
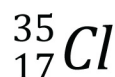
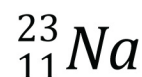


Practicing Protons, Neutrons, and Electrons

1.

Answers

Element name: Helium# of protons: 2# of neutrons: 2Element name: Iron# of protons: 26# of neutrons: 30Element name: Aluminium# of protons: 26# of neutrons: 30Element name: Calcium# of protons: 20# of neutrons: 20Element name: Neon# of protons: 10# of neutrons: 10Element name: Carbon# of protons: 6# of neutrons: 6Element name: Fluorine# of protons: 9# of neutrons: 10Element name: Chlorine# of protons: 17# of neutrons: 18Element name: Sodium# of protons: 11# of neutrons: 12

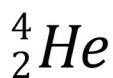
2.

Element	Atomic Number	Atomic Mass	Number of protons	Number of neutrons	Number of electrons
S	16	32	16	16	16
Cl	17	35	17	18	17
Ar	18	40	18	22	18
K	19	39	19	20	19
Ca	20	40	20	20	20

Name : _____ Score : _____ Date : _____

Practicing Protons, Neutrons, and Electrons

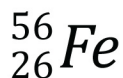
1. Fill in the element name and the number of protons and neutrons for the given symbols.



Element name: _____

of protons: _____

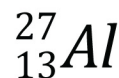
of neutrons: _____



Element name: _____

of protons: _____

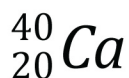
of neutrons: _____



Element name: _____

of protons: _____

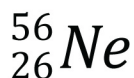
of neutrons: _____



Element name: _____

of protons: _____

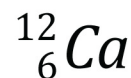
of neutrons: _____



Element name: _____

of protons: _____

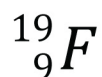
of neutrons: _____



Element name: _____

of protons: _____

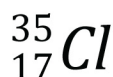
of neutrons: _____



Element name: _____

of protons: _____

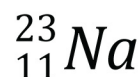
of neutrons: _____



Element name: _____

of protons: _____

of neutrons: _____



Element name: _____

of protons: _____

of neutrons: _____

2. Using the periodic table, fill in the missing information for the given elements.

Element	Atomic Number	Atomic Mass	Number of protons	Number of neutrons	Number of electrons
S	16		16		
	17	35			17
	18		18	22	
	19	39			19
Ca	20		20		