

Ionic Compound Formula Worksheet

1. Write the formula for the following compounds:

a) Ammonium Dichromate →

b) Chromium (III) Nitrite →

c) Lead(II) Oxide →

d) Lithium Hydrogen Sulfate →

e) Tin(IV) Bromide →

2. Name these ionic compounds:

a) CaSO_4 →

b) KClO →

c) KMnO_4 →

d) LiCN →

e) $\text{Fe}(\text{ClO}_3)_3$ →

3. Write the name & formulas for ionic compounds formed from these pairs of ions:

a) NH_4^+ , SO_3^{2-} →

b) Ca^{2+} , PO_4^{3-} →

c) Al^{3+} , NO_3^- →

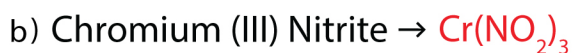
d) K^+ , CrO_4^{2-} →

e) Sr^{2+} , F^- →

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1.

Answers



2.



3.

