

Name : \_\_\_\_\_

# Ionic Compounds: Names and Formulas Worksheet

Write the names of the compounds whose formulas have been given in the following table.

Formula	Name of the Compound
NaBr	
FeO	
LiI	
AgCN	
$(\text{NH}_4)_2\text{SO}_4$	
$\text{V}_3\text{N}_5$	
$\text{Cu}_2\text{CO}_3$	
$(\text{NH}_4)_2\text{O}$	
$\text{CoBr}_2$	
$\text{Mn}_3\text{As}_7$	
$\text{Ag}_3\text{PO}_4$	
$\text{Fe}_2\text{O}_3$	
$\text{Na}_3\text{P}$	
$\text{Cu}(\text{ClO}_3)_2$	
$\text{Cu}_3\text{As}$	
$\text{Fe}(\text{HCO}_3)_2$	
$\text{Mg}(\text{CH}_3\text{COO})_2$	
$\text{Co}_2(\text{CrO}_4)_3$	

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## Answers

Formula	Name of the Compound
NaBr	Sodium Bromide
FeO	Iron (II) Oxide
LiI	Lithium Iodide
AgCN	Silver Cyanide
$(\text{NH}_4)_2\text{SO}_4$	Ammonium Sulfate
$\text{V}_3\text{N}_5$	Vanadium (V) Nitride
$\text{Cu}_2\text{CO}_3$	Copper (I) Carbonate
$(\text{NH}_4)_2\text{O}$	Ammonium Oxide
$\text{CoBr}_2$	Cobalt (II) Bromide
$\text{Mn}_3\text{As}_7$	Manganese (VII) Arsenide
$\text{Ag}_3\text{PO}_4$	Silver Phosphate
$\text{Fe}_2\text{O}_3$	Iron (III) Oxide
$\text{Na}_3\text{P}$	Sodium Phosphide
$\text{Cu}(\text{ClO}_3)_2$	Copper (II) Chlorate
$\text{Cu}_3\text{As}$	Copper (I) Arsenide
$\text{Fe}(\text{HCO}_3)_2$	Iron (II) Bicarbonate
$\text{Mg}(\text{CH}_3\text{COO})_2$	Magnesium Acetate
$\text{Co}_2(\text{CrO}_4)_3$	Cobalt (III) Chromate