

Ionic Compounds Practice Worksheet



Write the formulas of these ionic compounds.

1. Calcium Bromide →
2. Copper(I) Oxide →
3. Manganese (VII) Arsenide →
4. Ammonium Oxide →
5. Silver Cyanide →
6. Lithium Iodide →
7. Scandium Hydroxide →
8. Vanadium(III) Sulfide →
9. Iron (II) Bromide →
10. Gallium Nitride →
11. Vanadium (V) Phosphide →
12. Beryllium Hydroxide →
13. Magnesium Acetate →
14. Tin (II) Nitrite →
15. Titanium (II) Selenide →
16. Potassium Carbonate →
17. Cobalt (III) Chlorate →
18. Beryllium Acetate →

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Answers

1. Calcium Bromide → CaBr_2
2. Copper(I) Oxide → Cu_2O
3. Manganese (VII) Arsenide → Mn_3As_7
4. Ammonium Oxide → $(\text{NH}_4)_2\text{O}$
5. Silver Cyanide → AgCN
6. Lithium Iodide → LiI
7. Scandium Hydroxide → $\text{Sc}(\text{OH})_3$
8. Vanadium(III) Sulfide → V_2S_3
9. Iron (II) Bromide → FeBr_2
10. Gallium Nitride → GaN
11. Vanadium (V) Phosphide → V_3P_5
12. Beryllium Hydroxide → $\text{Be}(\text{OH})_2$
13. Magnesium Acetate → Mn_3As_7
14. Tin (II) Nitrite → $\text{Sn}(\text{NO}_2)_2$
15. Titanium (II) Selenide → TiSe
16. Potassium Carbonate → K_2CO_3
17. Cobalt (III) Chlorate → $\text{Co}(\text{ClO}_3)_3$
18. Beryllium Acetate → $\text{Be}(\text{CH}_3\text{COO})_2$