

Name : \_\_\_\_\_

# Transition Metals Ionic Compounds Worksheet

Write the formulae of the following compounds.

<b>Name of the Compound</b>	<b>Formulas</b>
Zinc Nitrate	
Copper (I) Oxide	
Silver Bromide	
Cobalt (II) Hydroxide	
Lead (IV) Sulfate	
Iron (II) Chloride	
Silver Nitrate	
Copper (I) Chlorate	
Chromium (III) Hydroxide	
Mercury (I) Oxide	
Chromium (III) Cyanide	
Zinc Cyanide	
Lead (II) Sulfite	
Cobalt (III) Carbonate	

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## Answers

Name of the Compound	Formulas
Zinc Nitrate	$\text{Zn}(\text{NO}_3)_2$
Copper (I) Oxide	$\text{Cu}_2\text{O}$
Silver Bromide	$\text{AgBr}$
Cobalt (II) Hydroxide	$\text{Co}(\text{OH})_2$
Lead (IV) Sulfate	$\text{Pb}(\text{SO}_4)_2$
Iron (II) Chloride	$\text{FeCl}_2$
Silver Nitrate	$\text{AgNO}_3$
Copper (I) Chlorate	$\text{CuClO}_3$
Chromium (III) Hydroxide	$\text{Cr}(\text{OH})_3$
Mercury (I) Oxide	$\text{Hg}_2\text{O}$
Chromium (III) Cyanide	$\text{Cr}(\text{CN})_3$
Zinc Cyanide	$\text{Zn}(\text{CN})_2$
Lead (II) Sulfite	$\text{PbSO}_3$
Cobalt (III) Carbonate	$\text{Co}_2(\text{CO}_3)_3$