

SIMPLE BINARY IONIC COMPOUNDS WORKSHEET

Fill the table with the correct information.

Name of the Compound	Formulas
Iron (II) Chloride	
Copper (II) Sulfate	
Zinc Nitrate	
Silver Bromide	
Iron (III) Acetate	
Lead (II) Sulfite	
Cobalt (III) Carbonate	
Copper (I) Oxide	
Silver Nitrate	
Copper (I) Chlorate	
Mercury (I) Oxide	
Chromium (III) Cyanide	
Chromium (III) Hydroxide	
Cobalt (II) Hydroxide	
Zinc Hydroxide	

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Answers

Name of the Compound	Formulas
Iron (II) Chloride	FeCl_2
Copper (II) Sulfate	CuSO_4
Zinc Nitrate	$\text{Zn}(\text{NO}_3)_2$
Silver Bromide	AgBr
Iron (III) Acetate	$\text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_2$
Lead (II) Sulfite	PbSO_3
Cobalt (III) Carbonate	$\text{Co}_2(\text{CO}_3)_3$
Copper (I) Oxide	Cu_2O
Silver Nitrate	AgNO_3
Copper (I) Chlorate	CuClO_3
Mercury (I) Oxide	Hg_2O
Chromium (III) Cyanide	$\text{Cr}(\text{CN})_3$
Chromium (III) Hydroxide	$\text{Cr}(\text{OH})_3$
Cobalt (II) Hydroxide	$\text{Co}(\text{OH})_2$
Zinc Hydroxide	$\text{Zn}(\text{OH})_2$