

TYPE 2 BINARY IONIC COMPOUNDS WORKSHEET

Write the formulas of the following Type 2 binary ionic compounds.

Copper (II) Iodide		Copper (I) Oxide	
Iron (II) Sulfide		Manganese (IV) Oxide	
Gold (III) Chloride		Gold (I) Sulfide	
Silver Bromide		Cadmium Sulfide	
Vanadium (V) Nitride		Chromium (VI) Oxide	
Lead (IV) Sulfide		Zinc Selenide	
Gallium (I) Iodide		Vanadium (III) Chloride	
Nickel (III) Selenide		Iron (III) Bromide	
Silver Sulfide		Tin (IV) Chloride	
Manganese (II) Fluoride		Antimony (V) Bromide	
Cobalt (III) Oxide		Titanium (IV) Oxide	

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Write the formulas of the following Type 2 binary ionic compounds.

Answers

Copper (II) Iodide	CuI_2	Copper (I) Oxide	Cu_2O
Iron (II) Sulfide	FeS	Manganese (IV) Oxide	MnO_2
Gold (III) Chloride	AuCl_3	Gold (I) Sulfide	Au_2S
Silver Bromide	AgBr	Cadmium Sulfide	CdS
Vanadium (V) Nitride	V_3N_5	Chromium (VI) Oxide	CrO_3
Lead (IV) Sulfide	PbS_2	Zinc Selenide	ZnSe
Gallium (I) Iodide	GaI	Vanadium (III) Chloride	VCl_3
Nickel (III) Selenide	Ni_2Se_3	Iron (III) Bromide	FeBr_3
Silver Sulfide	Ag_2S	Tin (IV) Chloride	SnCl_4
Manganese (II) Fluoride	MnF_2	Antimony (V) Bromide	SbBr_5
Cobalt (III) Oxide	Co_2O_3	Titanium (IV) Oxide	TiO_2