

Name : Date :

Drawing Ionic and Covalent Bonds Worksheet

Draw the Lewis Dot Structures of the compounds formed by the following atoms and mention whether the bonds formed by them are covalent or ionic.

1. Sodium and Chlorine

2. Hydrogen and Oxygen

3. Chlorine and Chlorine

4. Lithium and Bromine

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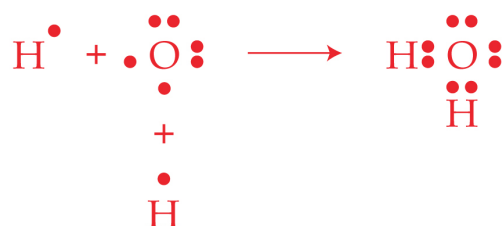
Draw the Lewis Dot Structures of the compounds formed by the following atoms and mention whether the bonds formed by them are covalent or ionic.

1. Sodium and Chlorine



The bond is ionic.

2. Hydrogen and Oxygen



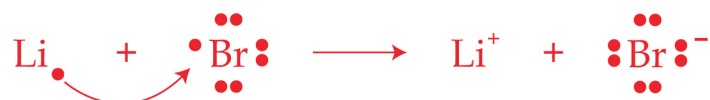
The bonds are polar covalent.

3. Chlorine and Chlorine



The bond formed is non-polar covalent.

4. Lithium and Bromine



The bond formed is ionic.