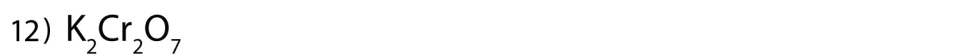
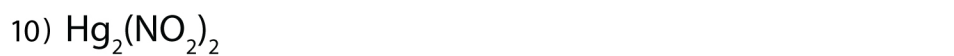
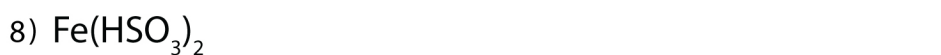


Naming Polyatomic Ionic Compounds Worksheet

Name the following ions.



Naming Polyatomic Ionic Compounds Worksheet

Answers

- 1) $(\text{NH}_4)_2\text{CrO}_4$ Ammonium Chromate
- 2) AgNO_3 Silver Nitrate
- 3) $\text{Al}(\text{ClO}_3)_3$ Aluminum Chlorate
- 4) $\text{Ba}(\text{ClO}_4)_2$ Barium Perchlorate
- 5) CaSO_4 Calcium Sulfate
- 6) $\text{Cd}(\text{OH})_2$ Cadmium Hydroxide
- 7) $\text{Cu}(\text{ClO}_4)_2$ Copper (II) Chlorate
- 8) $\text{Fe}(\text{HSO}_3)_2$ Iron (II) Bisulfite
- 9) $\text{Fe}(\text{NO}_2)_2$ Iron (II) Nitrite
- 10) $\text{Hg}_2(\text{NO}_2)_2$ Mercury (I) Nitrite
- 11) HgCrO_4 Mercury (II) Chromate
- 12) $\text{K}_2\text{Cr}_2\text{O}_7$ Potassium Dichromate
- 13) $\text{Mg}_3(\text{PO}_4)_2$ Magnesium Phosphate
- 14) $\text{Pb}(\text{CO}_3)_2$ Lead (II) Carbonate
- 15) $\text{Pb}(\text{SO}_4)_2$ Lead (IV) Sulfate
- 16) $\text{Pb}_3(\text{PO}_4)_2$ Lead (II) Phosphate
- 17) $\text{Sn}(\text{CN})_2$ Tin (II) Cyanide
- 18) $\text{Sn}(\text{CO}_3)_2$ Tin (IV) Carbonate
- 19) $\text{Sr}(\text{NO}_3)_2$ Strontium Nitrate
- 20) $\text{Zn}(\text{HCO}_3)_2$ Zinc Bicarbonate