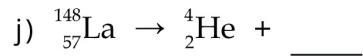
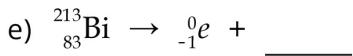
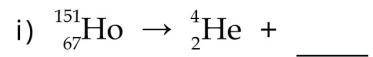
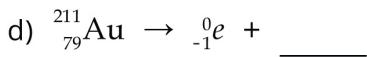
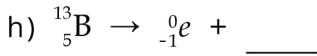
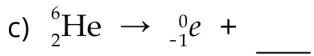
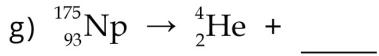
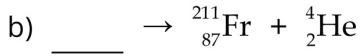
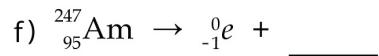
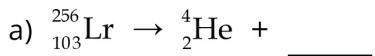


Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Nuclear Equation Worksheet

1. Complete the following nuclear decay by filling in the missing nuclei.



2. Write equations for the following nuclear reactions.

a) Decay of polonium – 218 by alpha ( $\alpha$ ) emission

b) Decay of carbon – 14 by beta ( $\beta$ ) emission

c) Alpha decay of radon – 198

d) Beta decay of uranium – 237

e) Beta decay of cobalt - 60

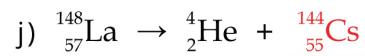
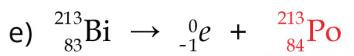
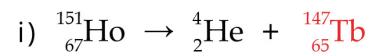
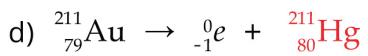
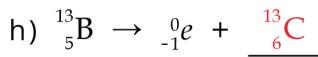
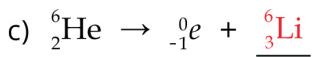
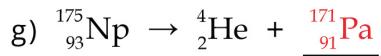
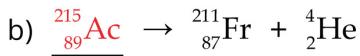
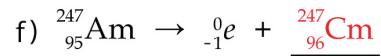
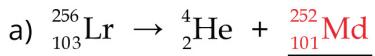
f) Positron emission of fluorine – 18

g) Electron capture of krypton – 89

Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Nuclear Equation Worksheet

1. Complete the following nuclear decay by filling in the missing nuclei.

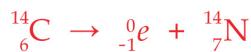


2. Write equations for the following nuclear reactions.

a) Decay of polonium – 218 by alpha ( $\alpha$ ) emission



b) Decay of carbon – 14 by beta ( $\beta$ ) emission



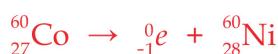
c) Alpha decay of radon – 198



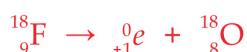
d) Beta decay of uranium – 237



e) Beta decay of cobalt - 60



f) Positron emission of fluorine – 18



g) Electron capture of krypton – 89

