

Polyatomic Ions Practice Worksheet

Complete the following table with the appropriate information.

Name of the Compound	Formula
Beryllium Hydroxide	
Barium Iodate	
Sodium Nitrite	
Gallium Cyanate	
Ammonium Chloride	
Silver Sulfite	
Calcium Bisulfate	
Magnesium Hypocarbonite	
Rubidium Perchlorate	
Ammonium Nitrite	
Strontium Sulfite	
Aluminum Acetate	
Strontium Persulfate	
Ammonium Nitrate	
Rubidium Chlorite	
Calcium Bisulfate	
Silver Hyposulfite	
Ammonium Chloride	

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Answers

Name of the Compound	Formula
Beryllium Hydroxide	$\text{Be}(\text{OH})_2$
Barium Iodate	$\text{Ba}(\text{IO}_3)_2$
Sodium Nitrite	NaNO_2
Gallium Cyanate	$\text{Ga}(\text{CNO})_3$
Ammonium Chloride	NH_4Cl
Silver Sulfite	Ag_2SO_3
Calcium Bisulfate	$\text{Ca}(\text{HSO}_4)_2$
Magnesium Hypocarbonite	MgCO
Rubidium Perchlorate	RbClO_4
Ammonium Nitrite	$(\text{NH}_4)\text{NO}_2$
Strontium Sulfite	SrSO_3
Aluminum Acetate	$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$
Strontium Persulfate	SrSO_5
Ammonium Nitrate	NH_4NO_3
Rubidium Chlorite	RbClO_2
Calcium Bisulfate	$\text{Ca}(\text{HSO}_4)_2$
Silver Hyposulfite	Ag_2SO_2
Ammonium Chloride	NH_4Cl