

# MOLAR MASS CALCULATION WORKSHEET

Calculate the molar masses of the following.

1. KOH

2. Cl<sub>2</sub>

3. NaCl

4. Fe<sub>2</sub>O<sub>3</sub>

5. FeCl<sub>3</sub>

6. SO<sub>2</sub>

7. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>

8. Na<sub>2</sub>O

# MOLAR MASS CALCULATION WORKSHEET

## Answers

1. KOH

Molar mass of KOH = Molar mass of K + Molar mass of O + Molar mass of H =  
 $39.1 + 16 + 1.01 = 56.11 \text{ g/mol}$

2. Cl<sub>2</sub>

Molar mass of Cl<sub>2</sub> = 2 x Molar mass of Cl = 71 g/mol

3. NaCl

Molar mass of NaCl = Molar mass of Na + Molar mass of Cl = 22.99 + 35.45 =  
58.44 g/mol

4. Fe<sub>2</sub>O<sub>3</sub>

Molar mass of Fe<sub>2</sub>O<sub>3</sub> = (2 x Molar mass of Fe) + (3 x Molar mass of O) = 111.7 + 48 =  
159.7 g/mol

5. FeCl<sub>3</sub>

Molar mass of FeCl<sub>3</sub> = Molar mass of Fe + (3 x Molar mass of Cl) = 55.85 + 106.35 =  
162.2 g/mol

6. SO<sub>2</sub>

Molar mass of SO<sub>2</sub> = Molar mass of S + (2 x Molar mass of O) = 32.065 + 32 =  
64.065 g/mol

7. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>

Molar mass of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> = (2 x Molar mass of N) + (8 x Molar mass of H) +  
Molar mass of S + (4 x Molar mass of O) = 28.02 + 8.08 + 64 = 100.1 g/mol

8. Na<sub>2</sub>O

Molar mass of Na<sub>2</sub>O = (2 x Molar mass of Na) + Molar mass of O = 46 + 16 = 62 g/mol